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ecules together in each type of solid. List your answers in

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column F in the table above. Read This! When a solid substance is heated, the weakest attractive forces in the solid structure are overcome, and the solid melts. In some cases, this results in individual atoms or ions, and in other cases, this results in individual molecules.

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(Your answer should include a discussion about concentration gradient.) The electrons go against the concentration gradient and then something happens THIS SET IS OFTEN IN FOLDERS WITH...

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Describe the trend in solubility for Substances A and B in Model 4 as temperature increases. 32. If a saturated solution of Substance A in 100.0 g of water is cooled from 300C to 100C, what mass of solid

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questions. 1. What are the customary units of solubility on solubility curves? Yemp 2. Define solubility. ok be aya 3. According to the graph, the solubility of any substance changes as changes. 4. List the substances whose solubility decreases as temperature increases. 5.

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4<sup>™</sup> Activities for AP\* Chemistry POGIL 14. The solubility product constant,  $K_{sp}$ , for silver carbonate,  $Ag_2CO_3$ , is  $8.10 \times 10^{-12}$ . Compare the solubility of silver carbonate in water to its

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solubility in a 0.100 M solution of silver nitrate. Support your comparison with calculations similar to those in Model 1.

### **Common Ion Effect on Solubility - science with ms. hall**

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## **Solubility Curve Lab Answer Key**

Solubility,  $K_{sp}$  Answer Key WORKSHEET 1 1) A 2) A 3) C 4) A 5) D 6) B 7) C 8) C 9) C 10) A 11) a) drives rxn to the left b) drives rxn to the right c) drives rxn to the right d) drives rxn to the left 12) C . Kinetics Worksheet 1 1. The rate of a chemical reaction between substances A and B is found to follow the rate equation

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