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Question: Second Test Semester Review
Sheet Part II Math Problems (Concluded
6) In the reaction above (Mg# 24.31;
O-16.00; H# 1.01; P:-30.97) A) 3.498×10 g of H₂PO₄ will require how many
moles of Mg(OH)₂ B) 3.498×10 g of H₃
PO₄ will produce how many molecules
of Mg₂(PO₄)₂ C) 3.498×10 g of H₃PO₄
will require how many grams of H₂O?
D) 9.876×10 molecules of ...

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