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Matter changes phase  
when energy is added  
or removed. freezing  
point condensation  
deposition phase  
diagram triple point  
SECTION 12.4 Phase  
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contrast sublimation  
and evaporation. In  
both processes, the  
substances become a  
vapor. During  
sublimation, the  
substance goes from

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the solid phase directly  
to the vapor phase.

During evaporation,  
particles in a liquid  
gain enough

## **States of Matter**

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Challenge Air is a  
mixture of gases By  
percentage, it is  
roughly 78 percent  
nitrogen, 21 percent  
oxygen, and 1 percent  
argon (There are trace  
amounts of many other  
gases in air) If the  
atmospheric pressure

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distefanoa17. States of  
Matter. Terms in this  
set (35) Kinetic  
Molecular Theory.  
describes the behavior  
of matter in terms of

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particles in motion.  
Elastic Collision.  
collision in which no  
kinetic energy is lost.

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motion 2. a. small b.  
forces c. random d.

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elastic; kinetic 3. KE  
 $\frac{1}{2} mv^2$  4.  
Temperature 5. true 6.  
true 7. false 8. true 9.  
true 10. false 11. true  
12. false 13. a 14. a 15.  
d 16. d 17. b 18. b 19.  
b 20. barometer 21.  
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Section 10.2 Mass and

the Mole • 1 mole is

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the amount of atoms in  
12 g of pure carbon- ...  
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