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7.56 g of zwitterionic form of cysteine (MW=121.16 g/mol) was dissolved in 400ml of 0.18 m HCl. A 25.00ml volume of this solution was then titrated with 0.489

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Question: Chemistry 112 Pre-lab  
Exercise Experiment 5 Exercise Number  
257 A) Using A First Order Plot,  
Determine The Rate Constant  $k_t$ , For  
The Oxidation Of Benzyl Alcohol With  
Potassium Permanganate At The  
Temperature,  $T$ , 20.2 C, From Below.

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(Molar Absorptivity,  $\epsilon = 2350 \text{ M}^{-1}\text{cm}^{-1}$ ,  
Path Length,  $L = 1 \text{ cm}$ ) The Information  
In The Table Time (minutes) Absorbance  
15 30 ...

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of Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub>, Molarity (M) 1/time (1/sec)

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